

## Base-catalyzed synthesis of 2-thioxo-2,3-dihydrothieno[2,3-d]pyrimidin-4(1H)-ones and isolation of intermediates using microwave irradiation

A. Davoodnia\*, M. Bakavoli, N. Zareei, N. Tavakoli-Hoseini

Department of Chemistry, School of Sciences, Islamic Azad University,  
Mashhad Branch, Mashhad 91735-413, Iran

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A simple and fast method for the synthesis of some 3-substituted-5,6-dimethyl-2-thioxo-2,3-dihydrothieno[2,3-d]pyrimidin-4(1H)-ones has been developed *via* base-catalyzed cyclocondensation of ethyl 2-amino-4,5-dimethylthiophene-3-carboxylate with isothiocyanates. The uncyclized intermediates, ethyl 4,5-dimethyl-2-[(substituted carbamothioyl)amino]thiophene-3-carboxylates, were isolated when the reactions were carried out under microwave irradiation. These intermediates subsequently underwent cyclization in *t*-butanol in the presence of potassium *t*-butoxide on heating under reflux to give the desired bicyclic products.

**Key words:** ethyl 2-amino-4,5-dimethylthiophene-3-carboxylate, isothiocyanates, 2-thioxo-2,3-dihydrothieno[2,3-d]pyrimidin-4(1H)-ones, microwave irradiation.

### INTRODUCTION

Our interest in thieno[2,3-d]pyrimidine synthesis emerges from the numerous reports on their diverse biological activities [1–10]. Various methods have already been proposed for the synthesis of these compounds and the most general ones involve cyclocondensation of suitably functionalized thiophenes with different electrophiles such as chloroformamide [11],  $\alpha$ -substituted acetonitriles [12], formic acid [13], phosgene [14], ethyl chloroformate [14] and guanidine [15]. To the best of our knowledge, base-catalyzed cyclocondensation of ethyl 2-amino-4,5-dimethylthiophene-3-carboxylate (**1**) with isothiocyanates for the synthesis of 3-substituted-5,6-dimethyl-2-thioxo-2,3-dihydrothieno[2,3-d]pyrimidin-4(1H)-ones (**3a–e**) and the utilization of microwave irradiation for isolation of the intermediates (**2a–e**) has not been reported in the literature.

Prompted by these findings and due to our interest in the synthesis of heterocyclic compounds [16–25] and in continuation of our previous works on the synthesis of thieno[2,3-d]pyrimidine derivatives [26–28], we report here a simple and fast method for the synthesis of 3-substituted-5,6-dimethyl-2-thioxo-2,3-dihydrothieno[2,3-d]pyrimidin-4(1H)-ones (**3a–e**) through cyclocondensation of ethyl 2-amino-4,5-dimethylthiophene-3-carboxylate (**1**) with isothiocyanates under basic conditions.

### RESULTS AND DISCUSSION

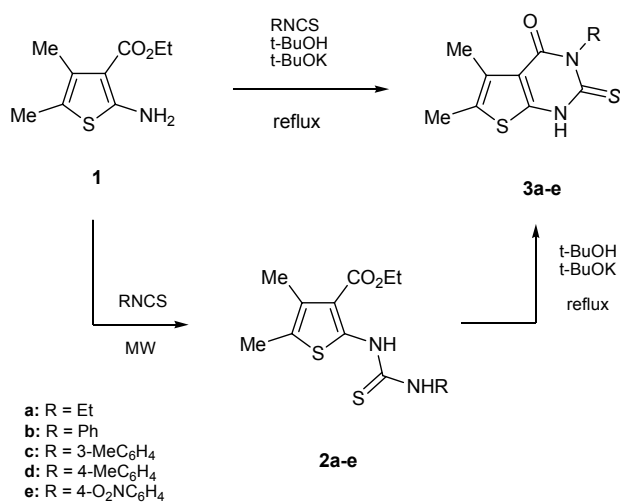
The starting material (**1**) was prepared according to the literature method [29]. Cyclocondensation of this compound with isothiocyanates in the presence of potassium *t*-butoxide in *t*-butanol under reflux gave products identified as 3-substituted-5,6-dimethyl-2-thioxo-2,3-dihydrothieno[2,3-d]pyrimidin-4(1H)-ones (**3a–e**). Under this conditions, attempts to isolate the reaction intermediates (**2a–e**) failed when we monitored the course of the reactions carefully (Scheme 1).

Due to our interest in the utilization of microwave irradiation for the synthesis of heterocyclic compounds [30–33], we tried to extend this non-conventional synthetic method for the synthesis of compounds **3a–e**. Therefore, ethyl 2-amino-4,5-dimethylthiophene-3-carboxylate (**1**) was allowed to interact with isothiocyanates under microwave irradiation under solvent-free conditions at 800 W. During monitoring of the reaction mixture by TLC (CHCl<sub>3</sub>:MeOH = 95:5), surprisingly, we observed that unexpected products, with R<sub>F</sub>-values different from those expected for compounds **3a–e**, were being formed. During work up and identification, it was established that a condensation and not a cyclocondensation reaction had occurred and the intermediates ethyl 4,5-dimethyl-2-[(substituted carbamothioyl)amino]thiophene-3-carboxylates (**2a–e**) were isolated. The reaction did not proceed to form cyclic products even after prolonged irradiation, but when the latter compounds were heated under reflux for 3 hours in the presence of potassium *t*-butoxide

\* To whom all correspondence should be sent:  
E-mail: adavoodnia@yahoo.com

in *t*-butanol, cyclization reaction occurred and the cyclic products **3a–e** were obtained (Scheme 1).

The structure of the synthesized compounds was deduced from their spectral and microanalytical data. For example, the  $^1\text{H}$  NMR spectrum of **2a** did not show the  $\text{NH}_2$  signal of the precursor **1** at  $\delta$  5.61 ppm, but instead of it showed two broad signals at  $\delta$  9.62 and 11.80 ppm belonging to the NH groups indicating the formation of compound **2a**. The IR spectrum showed the absorption bands at 1651, 3215 and 3299  $\text{cm}^{-1}$  for carbonyl and two NH groups respectively. The MS of **2a** showed a molecular ion peak at  $m/z$  286 ( $\text{M}^+$ ) corresponding to the molecular formula  $\text{C}_{12}\text{H}_{18}\text{N}_2\text{O}_2\text{S}_2$ . This compound gave also satisfactory elemental analysis data (See Experimental).



Scheme 1.

In conclusion, we have developed a facile method for the synthesis of 3-substituted-5,6-dimethyl-2-thioxo-2,3-dihydrothieno[2,3-d]pyrimidin-4(1H)-ones (**3a–e**) through cyclocondensation of ethyl 2-amino-4,5-dimethylthiophene-3-carboxylate (**1**) with isothiocyanates in *t*-butanol containing potassium *t*-butoxide as a base catalyst. Using microwave irradiation, the uncyclized intermediates **2a–e** were isolated.

## EXPERIMENTAL

Melting points were measured on a Stuart Model SMP3 melting point apparatus. The IR spectra were obtained on a 4300 Shimadzu spectrophotometer as KBr disks. The  $^1\text{H}$  NMR (100 MHz) spectra were recorded on a Bruker AC 100 spectrometer. The mass spectra were determined on a Shimadzu GCMS 17A instrument. Elemental analysis was performed on a Thermo Finnigan Flash EA micro-analyzer. Reactions were performed in a domestic microwave oven Model LG MS-543XD.

**Preparation of 3-substituted-5,6-dimethyl-2-thioxo-2,3-dihydrothieno[2,3-d]pyrimidin-4(1H)-ones (3a–e). General Procedure. Method A.** To a solution of the ethyl 2-amino-4,5-dimethylthiophene-3-carboxylate (**1**) (5 mmol) and potassium *t*-butoxide (2 mmol) in *t*-butanol (20 ml), the appropriate isothiocyanate (6 mmol) was added. The reaction mixture was heated under reflux for 6 hours. After the completion of the reaction (monitored by TLC,  $\text{CHCl}_3:\text{MeOH} = 93:7$ ), the solvent was evaporated in vacuum, the residue was dissolved in water (15 ml) and subsequently neutralized by 1 N HCl. The crude product was collected and recrystallized from ethanol to give compounds **3a–e** in 75, 77, 68, 86 and 74% yields, respectively.

**Method B.** A mixture of ethyl 4,5-dimethyl-2-[(substituted carbamothioyl)amino]thiophene-3-carboxylates (**2a–e**) (3 mmol) and potassium *t*-butoxide (1 mmol) in *t*-butanol (15 ml) was heated under reflux for 3 hours. After the completion of the reaction (monitored by TLC,  $\text{CHCl}_3:\text{MeOH} = 93:7$ ), the solvent was evaporated in vacuum, the residue was dissolved in water (15 ml) and subsequently neutralized by 1 N HCl. The crude product was collected and recrystallized from ethanol to give compounds **3a–e** in 79, 78, 73, 91 and 75% yields, respectively.

**3-Ethyl-5,6-dimethyl-2-thioxo-2,3-dihydrothieno[2,3-d]pyrimidin-4(1H)-one (3a).** M.p. 257–259°C;  $^1\text{H}$  NMR (DMSO-*d*<sub>6</sub>,  $\delta$  ppm): 1.17 (t, 3H, J = 7 Hz, CH<sub>3</sub>), 2.24 (s, 6H, 2CH<sub>3</sub>), 4.36 (q, 2H, J = 7 Hz, CH<sub>2</sub>), 13.45 (br, 1H, NH); IRS (KBr disc):  $\nu$  1684 (C=O), 3137  $\text{cm}^{-1}$  (NH); MS,  $m/z$ : 240 ( $\text{M}^+$ ); Analytically calculated for  $\text{C}_{10}\text{H}_{12}\text{N}_2\text{O}_2\text{S}_2$ : C 49.97; H 5.03; N 11.66; S 26.68. Found: C 50.24; H 5.21; N 11.35; S 26.91.

**5,6-Dimethyl-3-phenyl-2-thioxo-2,3-dihydrothieno[2,3-d]pyrimidin-4(1H)-one (3b).** M.p. 325–327°C;  $^1\text{H}$  NMR (DMSO-*d*<sub>6</sub>,  $\delta$  ppm): 2.24 (s, 6H, 2CH<sub>3</sub>), 7.0–7.6 (m, 5H, phenyl), 13.65 (s br, 1H, NH); IRS (KBr disc):  $\nu$  1703 (C=O), 3152  $\text{cm}^{-1}$  (NH); MS,  $m/z$ : 288 ( $\text{M}^+$ ); Analytically calculated for  $\text{C}_{14}\text{H}_{12}\text{N}_2\text{O}_2\text{S}_2$ : C 58.31; H 4.19; N 9.71; S 22.24. Found: C 58.67; H 3.98; N 9.50; S 22.45.

**5,6-Dimethyl-3-(3-methylphenyl)-2-thioxo-2,3-dihydrothieno[2,3-d]pyrimidin-4(1H)-one (3c).** M.p. 295–297°C;  $^1\text{H}$  NMR (DMSO-*d*<sub>6</sub>,  $\delta$  ppm): 2.25 (s, 3H, CH<sub>3</sub>), 2.29 (s, 6H, 2CH<sub>3</sub>), 6.85–7.45 (m, 4H, arom-H), 13.61 (br, 1H, NH); IRS (KBr disc):  $\nu$  1709 (C=O), 3163  $\text{cm}^{-1}$  (NH); MS,  $m/z$ : 302 ( $\text{M}^+$ ); Analytically calculated for  $\text{C}_{15}\text{H}_{14}\text{N}_2\text{O}_2\text{S}_2$ : C 59.57; H 4.67; N 9.26; S 21.21. Found: C 59.28; H 4.89; N 9.51; S 20.97.

**5,6-Dimethyl-3-(4-methylphenyl)-2-thioxo-2,3-dihydrothieno[2,3-d]pyrimidin-4(1H)-one (3d).** M.p.

270°C (dec); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, δ ppm): 2.24 (s, 3H, CH<sub>3</sub>), 2.28 (s, 3H, CH<sub>3</sub>), 2.34 (s, 3H, CH<sub>3</sub>), 6.90–7.25 (overlapped doublets, 4H, arom-H), 13.60 (br, 1H, NH); IRS (KBr disc): ν 1703 (C=O), 3155 cm<sup>-1</sup> (NH); MS, m/z: 302 (M<sup>+</sup>); Analytically calculated for C<sub>15</sub>H<sub>14</sub>N<sub>2</sub>O<sub>2</sub>S<sub>2</sub>: C 59.57; H 4.67; N 9.26; S 21.21. Found: C 59.34; H 4.46; N 9.48; S 21.43.

*5,6-Dimethyl-3-(4-nitrophenyl)-2-thioxo-2,3-dihydrothieno[2,3-d]pyrimidin-4(1H)-one (3e)*. M.p. 340–342°C; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, δ ppm): 2.25 (s, 3H, CH<sub>3</sub>), 2.32 (s, 3H, CH<sub>3</sub>), 7.40–8.45 (overlapped doublets, 4H, arom-H), 13.68 (s br, 1H, NH); IRS (KBr disc): ν 1707 (C=O), 3170 cm<sup>-1</sup> (NH); MS, m/z: 333 (M<sup>+</sup>); Analytically calculated for C<sub>14</sub>H<sub>11</sub>N<sub>3</sub>O<sub>3</sub>S<sub>2</sub>: C 50.44; H 3.33; N 12.60; S 19.24. Found: C 50.71; H 3.62; N 12.73; S 19.01.

*Preparation of ethyl 4,5-dimethyl-2-[(substituted carbamothioyl)amino]thiophene-3-carboxylates (2a–e)*. *General Procedure*. A mixture of ethyl 2-amino-4,5-dimethylthiophene-3-carboxylate (**1**) (3 mmol) and the appropriate isothiocyanate (4 mmol) was subjected to microwave irradiation at 800 W for 2–3 min (4–5 times). After the completion of the reaction (monitored by TLC, CHCl<sub>3</sub>:MeOH = 95:5) the crude product was recrystallized from ethanol to give compounds **2a–e** in high yields.

*Ethyl 2-[(ethylcarbamothioyl)amino]-4,5-dimethylthiophene-3-carboxylate (2a)*. Time 5×2 min; Yield 75%; m.p. 165–167°C; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, δ ppm): 1.44 (t, 3H, J = 6.5 Hz, CH<sub>3</sub>), 1.58 (t, 3H, J = 7 Hz, CH<sub>3</sub>), 2.32 (s, 6H, 2CH<sub>3</sub>), 4.41 (q, 2H, J = 6.5 Hz, CH<sub>2</sub>), 4.52 (q, 2H, J = 7 Hz, CH<sub>2</sub>), 9.62 (br, 1H, NH), 11.80 (br, 1H, NH); IRS (KBr disc): ν 1651 (C=O), 3215, 3299 cm<sup>-1</sup> (two NH); MS, m/z: 286 (M<sup>+</sup>); Analytically calculated for C<sub>12</sub>H<sub>18</sub>N<sub>2</sub>O<sub>2</sub>S<sub>2</sub>: C 50.32; H 6.33; N 9.78; S 22.39. Found: C 50.05; H 6.11; N 10.06; S 22.21.

*Ethyl 4,5-dimethyl 2-[(phenylcarbamothioyl)amino]thiophene-3-carboxylate (2b)*. Time 4×3 min; Yield 90%; m.p. 170–172°C; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, δ ppm): 1.22 (t, 3H, J = 7 Hz, CH<sub>3</sub>), 2.17 (s, 6H, 2CH<sub>3</sub>), 4.20 (q, 2H, J = 7 Hz, CH<sub>2</sub>), 7.0–7.6 (m, 5H, phenyl), 10.91 (s, 1H, NH), 11.79 (s, 1H, NH); IRS (KBr disc): ν 1664 (C=O), 3175, 3282 cm<sup>-1</sup> (two NH); MS, m/z: 334 (M<sup>+</sup>); Analytically calculated for C<sub>16</sub>H<sub>18</sub>N<sub>2</sub>O<sub>2</sub>S<sub>2</sub>: C 57.46; H 5.42; N 8.38; S 19.17. Found: C 57.78; H 5.19; N 8.64; S 18.98.

*Ethyl 4,5-dimethyl-2-[(3-methylphenyl)carbamothioyl]amino}thiophene-3-carboxylate (2c)*. Time 5×3 min; Yield 84%; m.p. 168–169°C; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, δ ppm): 1.48 (t, 3H, J = 7.5 Hz, CH<sub>3</sub>), 2.30 (s, 6H, 2CH<sub>3</sub>), 2.47 (s, 3H, CH<sub>3</sub>), 4.37 (q, 2H, J = 7.5 Hz, CH<sub>2</sub>), 7.1–7.6 (m, 4H, arom-H),

11.12 (s, 1H, NH), 12.10 (s, 1H, NH); IRS (KBr disc): ν 1659 (C=O), 3165, 3180 cm<sup>-1</sup> (two NH); MS, m/z: 348 (M<sup>+</sup>); Analytically calculated for C<sub>17</sub>H<sub>20</sub>N<sub>2</sub>O<sub>2</sub>S<sub>2</sub>: C 58.59; H 5.78; N 8.04; S 18.40. Found: C 58.94; H 6.01; N 7.81; S 18.59.

*Ethyl 4,5-dimethyl 2-[(4-methylphenyl)carbamothioyl]amino}thiophene-3-carboxylate (2d)*. Time 4×3 min; Yield 89%; m.p. 165–167°C; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, δ ppm): 1.25 (t, 3H, J = 7 Hz, CH<sub>3</sub>), 2.18 (s, 6H, 2CH<sub>3</sub>), 2.29 (s, 3H, CH<sub>3</sub>), 4.11 (q, 2H, J = 7 Hz, CH<sub>2</sub>), 7.1–7.5 (overlapped doublets, 4H, arom-H), 10.89 (s, 1H, NH), 11.78 (s, 1H, NH); IRS (KBr disc): ν 1658 (C=O), 3178, 3200 cm<sup>-1</sup> (two NH); MS, m/z: 348 (M<sup>+</sup>); Analytically calculated for C<sub>17</sub>H<sub>20</sub>N<sub>2</sub>O<sub>2</sub>S<sub>2</sub>: C 58.59; H 5.78; N 8.04; S 18.40. Found: C 58.81; H 5.54; N 7.87; S 18.73.

*Ethyl 4,5-dimethyl 2-[(4-nitrophenyl)carbamothioyl]amino}thiophene-3-carboxylate (2e)*. Time 5×3 min; Yield 78%; m.p. 213–215°C; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, δ ppm): 1.26 (t, 3H, J = 7 Hz, CH<sub>3</sub>), 2.22 (s, 6H, 2CH<sub>3</sub>), 4.30 (q, 2H, J = 7 Hz, CH<sub>2</sub>), 7.7–8.4 (overlapped doublets, 4H, arom-H), 11.61 (s, 1H, NH), 12.08 (s, 1H, NH); IRS (KBr disc): ν 1654 (C=O), 3184, 3205 cm<sup>-1</sup> (two NH); MS, m/z: 379 (M<sup>+</sup>); Analytically calculated for C<sub>16</sub>H<sub>17</sub>N<sub>3</sub>O<sub>4</sub>S<sub>2</sub>: C 50.64; H 4.52; N 11.07; S 16.90. Found: C 50.35; H 4.77; N 10.79; S 16.62.

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СИНТЕЗА НА 2-ТИОКСО-2,3-ДИГИДРОТИЕНО[2,3-d]ПИРИМИДИН-4(1H)-ОНИ  
ЧРЕЗ БАЗИЧНА КАТАЛИЗА И ИЗОЛИРАНЕ НА МЕЖДИННИ СЪЕДИНЕНИЯ  
С ИЗПОЛЗВАНЕ НА МИКРОВЪЛНОВО ОБЛЪЧВАНЕ

А. Давудниа\*, М. Бакаволи, Н. Зариеи, Н. Таваколи-Хосейни

Департамент по химия, Училище по науки, Ислямски университет Азад, Отдел Маишад,  
Маишад 91735-413, Иран

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(Резюме)

Предложен е прост и бърз метод за синтезата на 3-заместени-5,6-диметил-2-тиоксо-2,3-дихидроотиено[2,3-d]пиримидин-4(1H)-они чрез циклокондензация на 2-амино-4,5-диметилтиофен-3-карбокситилат с изотиоцианати чрез базична катализа. Нециклизираните междинни продукти 4,5-диметил-2-[(заместени карбаматотиоил)амино]тиофен-3-карбоксилати са изолирани когато реакциите се провеждат при микровълново облъчване. Тези междинни продукти впоследствие циклизират в *t*-бутанол в присъствие на калиев *t*-бутоксид при нагряване с обратен хладник до получаване на желаните бициклични продукти.