

Synthesis and biological activity of 4-chloro-2-hydroxy-N-(5-methylene-4-oxo-2-aryl-thiazolidin-3-yl) benzamide

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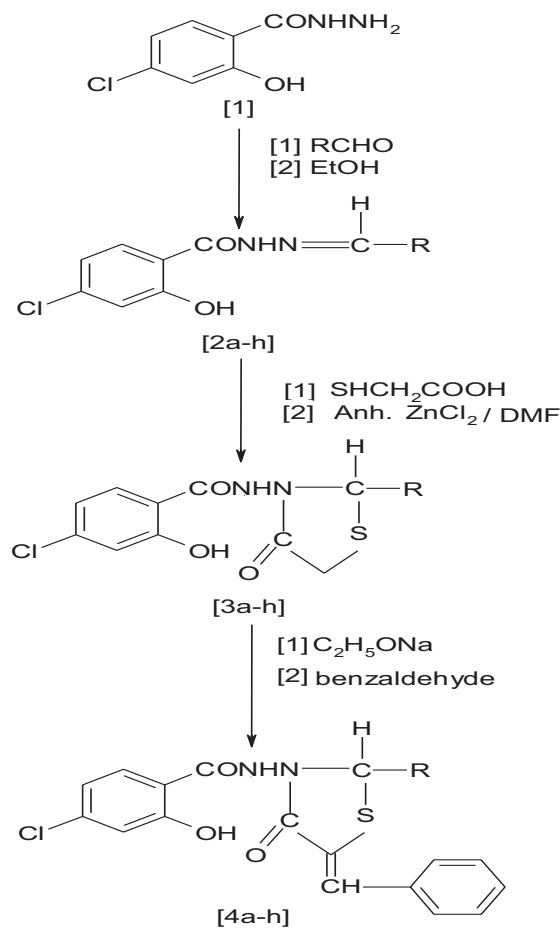
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4-chloro-2-hydroxy benzoic acid hydrazide (1) undergoes facile condensation with aromatic aldehydes to afford the corresponding 4-chloro-2-hydroxy benzoic acid arylidene hydrazides (2a-h) in good yields. Cyclocondensation of compounds (2a-h) with thioglycolic acid yields 4-chloro-2-hydroxy- N (4-oxo-2-aryl- thiazolidin -3-yl) benzamides (3a-h). These (3a-h) compounds are for the reacted with benzaldehyde in the presence of sodium ethanolate affords, giving 4-chloro-2-hydroxy- N (5-methylene-4-oxo-2-aryl- thiazolidin -3-yl)benzamides (4a-h). The structures of these compounds were established on the basis of analytical and spectral data. All the newly synthesized compounds were evaluated for their antibacterial and antifungal activities.

Key words: 4-chloro-2-hydroxy benzoic acid hydrazide, thiazolidin, antibacterial activity.

INTRODUCTION

Hydrazide and their heterocyclised products display diverse biological activities including antibacterial, antifungicidal, analgesic, anti-inflammatory properties [1–15]. These heterocyclic systems find wide use in medicine, agriculture and industry. One of the hydrazides, 4-chloro-2-hydroxy benzoic acid hydrazide and their condensed products play a vital role in medicinal chemistry [16–18]. 4-Thiazolidinones and its arylidene compounds give good pharmacological properties [19–23]. 4-thiazolidinones are also known to exhibit antitubercular [24], antibacterial [25], antifungal [26] and anticonvulsant activities. Hence, it was thought of interest to merge both of thiazolidinone and 4-chloro-2-hydroxy benzoic acid hydrazide moieties which may enhance the drug activity of compounds to some extent, or they might possess some of the above mentioned biological activities. From this point of view, the objective of the present work is to prepare new derivatives of 4-chloro-2-hydroxy benzoic acid hydrazide containing thiazolidinone moiety. Hence the present communication comprises the synthesis of 4-chloro-2-hydroxy- N (5-methylene-4-oxo-2-aryl- thiazolidin -3-yl) benzamide. The synthetic approach is shown in Scheme 1.



Scheme 1. R = (a) C₆H₅; (b) 4-OCH₃-C₆H₄; (c) 4-OH-C₆H₄; (d) 2-OH-C₆H₄; (e) 4-CH₃-C₆H₄; (f) 3,4-CH₂O₂-C₆H₄; (g) 4-OH-3-OCH₃-C₆H₃; (h) 3,4-C₂H₅-C₆H₄.

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EXPERIMENTAL

Melting points were determined in open capillary tubes and were not corrected. The IR spectra were recorded in KBr pellets on a Nicolet 400D spectrometer and ^1H NMR and ^{13}C NMR spectra were recorded in DMSO with TMS as internal standard on a Bruker spectrometer at 400 MHz and 100 MHz, respectively. LC-MS of selected samples taken on LC-MSD-Trap-SL_01046.

Preparation of 4-chloro-2-hydroxy benzoic acid arylidene hydrazide (2a-h)

General procedure: An equimolecular mixture of 4-chloro-2-hydroxy benzoic acid hydrazide (1), (0.01mole) and the aromatic aldehydes (a-h) in ethanol (15mL) was refluxed on a water bath for 1-2 h. The solid separated was collected by filtration, dried and recrystallized from ethanol. The yields, melting points and other characterization data of these compounds are given in Table 1.

Table .1 Analytical data and elemental analysis of compounds (2a-h)

Compd.	Molecular formula (Mol.wt.)	Yield	M.P.* °C	Elemental Analysis					
				%C		%H		%N	
				Found	Calcd.	Found	Calcd.	Found	Calcd.
2a	$\text{C}_{14}\text{H}_{11}\text{ClN}_2\text{O}_2$ (274)	85	243	61.18	61.21	3.99	4.04	10.15	10.20
2b	$\text{C}_{15}\text{H}_{13}\text{ClN}_2\text{O}_3$ (304)	80	246	59.08	59.12	4.25	4.30	9.14	9.19
2c	$\text{C}_{14}\text{H}_{11}\text{ClN}_2\text{O}_3$ (290)	75	240	57.79	57.84	3.77	3.81	9.58	9.64
2d	$\text{C}_{14}\text{H}_{11}\text{ClN}_2\text{O}_3$ (290)	81	243	57.78	57.84	3.75	3.81	9.57	9.64
2e	$\text{C}_{15}\text{H}_{13}\text{ClN}_2\text{O}_2$ (288)	79	244	62.36	62.40	4.51	4.54	9.64	9.70
2f	$\text{C}_{15}\text{H}_{11}\text{ClN}_2\text{O}_4$ (318)	75	247	56.49	56.53	3.44	3.48	8.73	8.79
2g	$\text{C}_{15}\text{H}_{13}\text{ClN}_2\text{O}_3$ (320)	77	249	56.14	56.17	3.04	4.09	8.68	8.73
2h	$\text{C}_{18}\text{H}_{19}\text{ClN}_2\text{O}_4$ (362)	73	261	59.55	59.59	5.24	5.28	7.68	7.72

* Uncorrected

Table 2. Spectral Data of Compounds (2a-h)

Compound	^1H NMR (δ , ppm)							
	Ar-H	-OH	-CONH	-N=CH	-CH ₃	-OCH ₃	-OC ₂ H ₅	-OCH ₂ O-cyclic
2a	6.85-7.84 (m, 8H)	11.70 (s)	11.86 (s)	8.40 (s)	-	-	-	-
2b	6.85-7.80 (m, 8H)	11.79 (s)	11.89 (s)	8.44 (s)	-	3.89 (3H)(s)	-	-
2c	6.88-7.84 (m, 8H)	11.85 (s)	11.88 (s)	8.47 (s)	-	-	-	-
2d	6.86-7.83 (m, 8H)	11.81 (s)	11.84 (s)	8.79 (s)	-	-	-	-
2e	6.85-7.80 (m, 8H)	11.80 (s)	11.91 (s)	8.46 (s)	2.39 (3H) (s)	-	-	-
2f	6.89-7.81 (m, 7H)	11.81 (s)	11.94 (s)	8.45 (s)	-	-	-	6.12, (2H) (s)
2g	6.87-7.84 (m, 7H)	11.76 (s)	11.88 (s)	8.47 (s)	-	3.89 (3H) (s)	-	-
2h	6.88-7.83 (m, 7H)	11.83 (s)	11.98 (s)	8.48(s)	-	-	4.12, 4H, (q) (CH ₂) 1.34, 6H, (t) (CH ₃)	-

Table 3. Spectral Data of Compounds (2a-h).

Compound	Ar-H	¹³ C NMR (δ, ppm)					
		-CONH	-N=CH	-OCH ₃	-OC ₂ H ₅	-CH ₃	-OCH ₂ O-cyclic
2a	112.2 -159.6	163.3	146.5	-	-	-	-
2b	113.8 - 159.5	163.4	146.3	55.7	-	-	-
2c	112.2 - 159.2	163.7	146.7	-	-	-	-
2d	112.7 - 159.2	163.9	146.5	-	-	-	-
2e	112.8 - 159.1	163.4	146.9	-	-	21.0	-
2f	113.3 - 160.8	163.6	146.7	-	-	-	102.5
2g	111.9 - 160.1	163-164	147.0	56.8	-	-	-
2h	112 - 160.5	163.8	146.9	-	65.4 (OCH ₂) 15.2 (CH ₃)	-	-

Table:-4 Analytical Data and Elemental Analysis of Compounds (3a-h).

Compd	Molecular formula (Mol.wt.)	Yield	M.P.* °C	Elemental Analysis							
				%C		%H		%N		%S	
				Found	Calcd.	Found	Calcd.	Found	Calcd.	Found	Calcd.
3a	C ₁₆ H ₁₃ ClN ₂ O ₃ S (348)	62	218	55.04	55.09	3.64	3.67	7.98	8.03	9.11	9.19
3b	C ₁₇ H ₁₅ ClN ₂ O ₄ S (344)	60	221	53.83	53.90	3.90	3.99	7.32	7.39	8.39	8.46
3c	C ₁₆ H ₁₃ ClN ₂ O ₄ S (364)	59	169	52.64	52.68	3.53	3.59	7.63	7.68	8.74	8.79
3d	C ₁₆ H ₁₃ ClN ₂ O ₄ S (364)	58	147	52.63	52.68	4.54	4.59	7.61	7.68	8.72	8.79
3e	C ₁₇ H ₁₅ ClN ₂ O ₃ S (362)	63	174	56.23	56.27	4.13	4.17	7.68	7.72	8.79	8.84
3f	C ₁₇ H ₁₃ ClN ₂ O ₅ S (392)	57	187	51.92	51.98	3.29	3.34	7.09	7.13	8.13	8.16
3g	C ₁₇ H ₁₅ ClN ₂ O ₅ S (394)	52	162	51.66	51.71	3.79	3.83	7.03	7.10	8.08	8.12
3h	C ₂₀ H ₂₁ ClN ₂ O ₅ S (436)	59	212	54.89	54.98	4.80	4.84	6.38	6.41	7.29	7.34

* Uncorrected

Preparation of 4-chloro-2-hydroxy- N-(4-oxo-2-aryl- thiazolidin -3-yl)benzamide (3a-h)

General procedure: A mixture 4-chloro-2-hydroxy benzoic acid arylidene hydrazide (2a-h) (0.01 mole) in THF (30mL) and mercapto acetic acid (thioglycolic acid) (0.01 mole) with a pinch of anhydrous ZnCl₂ was refluxed for 12 h. The solvent was then removed to get a residue, which was dissolved in benzene and passed through a column of silica gel using benzene: chloroform (8:2; v/v) mixture as eluent. The eluate was concentrated and

the product crystallized from alcohol to give 4-thiazolidinones (3a-h), which were obtained in 50–60% yield. The yields, melting points and other characterization data of these compounds are given in Table 4.

Preparation of 4-chloro-2-hydroxy- N (5-methylene-4-oxo-2-aryl- thiazolidin -3-yl) benzamide (4a-h)

Table 5. Spectral Data of Compounds (3a-h)

Compd.	¹ H NMR (δ, ppm)								
	Ar-H	-CH ₂ of the ring	-OH	-CH	-CONH	-CH ₃	-OCH ₃	-OC ₂ H ₅	-OCH ₂ O-cyclic
3a	6.70-7.85 (m, 8H)	3.85 (2H) (s)	11.72 (s)	5.90 (S)	11.84 (s)	-	-	-	-
3b	6.72-7.84 (m, 8H)	3.89 (2H) (s)	11.80 (s)	5.94 (S)	11.90 (s)	-	3.90 (3H)(s)	-	-
3c	6.73-7.85 (m, 8H)	3.86 (2H) (s)	11.83 (s)	5.92 (S)	11.88 (s)	-	-	-	-
3d	6.70-7.85 (m, 8H)	3.89 (2H) (s)	11.84 (s)	5.95 (S)	11.86 (s)	-	-	-	-
3e	6.73-7.84 (m, 8H)	3.90 (2H) (s)	11.78 (s)	5.94 (S)	11.98 (s)	2.40 (3H) (s)	-	-	-
3f	6.73-7.85 (m, 7H)	3.88 (2H) (s)	11.83 (s)	5.90 (S)	11.96 (s)	-	-	-	6.10, (2H) (s)
3g	6.72-7.85 (m, 7H)	3.91 (2H) (s)	11.74 (s)	5.93 (S)	11.90 (s)	-	3.90 (3H) (s)	-	-
3h	6.71-7.85 (m, 7H)	3.95 (2H) (s)	11.85 (s)	5.91 (S)	11.95 (s)	-	-	4.10, 4H, (q) (CH ₂) 1.34, 6H, (t) (CH ₃)	-

Table 6. Spectral Data of Compounds (3a-h)

Compd.	¹³ C NMR (δ, ppm)								
	Ar-H	-CH ₂	-CONH	-CH	-CO	-OCH ₃	-OC ₂ H ₅	-CH ₃	-OCH ₂ O-cyclic
3a	112.4-160.5	35.5	165.2	64.5	168.8	-	-	-	-
3b	112.6-160.8	35.4	164.8	64.8	168.9	55.8	-	-	-
3c	112.3-160.5	35.8	164.9	64.6	168.5	-	-	-	-
3d	112.8-159.8	35.6	165.1	64.5	168.7	-	-	-	-
3e	112.5-160.6	35.5	165.2	64.3	168.6	-	-	21.5	-
3f	112.3-160.7	35.8	164.5	64.2	168.9	-	-	-	102.8
3g	112.8-160.8	35.6	165.0	64.0	168.4	55.7	-	-	-
3h	113.0-160.5	35.4	164.8	64.7	169.0	-	65.1(OCH ₂) 14.8 (CH ₃)	-	-

Table 7. Analytical Data and Elemental Analysis of Compounds (4a-h)

Compd.	Molecular formula (Mol.wt.)	Yield	M.P. °C	Elemental Analysis							
				%C		%H		%N		%S	
				Found	Calcd.	Found	Calcd.	Found	Calcd.	Found	Calcd.
4a	C ₁₇ H ₁₃ ClN ₂ O ₃ S (360)	65	225	56.55	56.59	3.59	3.63	7.68	7.76	8.85	8.89
4b	C ₁₈ H ₁₅ ClN ₂ O ₄ S (390)	60	228	55.20	55.31	3.81	3.87	7.10	7.17	8.12	8.20
4c	C ₁₇ H ₁₃ ClN ₂ O ₄ S (376)	65	212	54.12	54.19	3.44	3.48	7.39	7.43	8.45	8.51
4d	C ₁₇ H ₁₃ ClN ₂ O ₄ S (376)	66	215	54.13	54.19	3.43	3.48	7.23	7.43	8.41	8.51
4e	C ₁₈ H ₁₅ ClN ₂ O ₃ S (374)	54	214	57.52	57.68	4.98	4.03	7.38	7.47	8.49	8.55
4f	C ₁₈ H ₁₃ ClN ₂ O ₅ S (404)	57	218	53.32	53.40	3.18	3.24	6.82	6.92	7.86	7.92
4g	C ₁₈ H ₁₅ ClN ₂ O ₅ S (406)	60	215	53.08	53.14	3.68	3.72	6.85	6.89	7.82	7.88
4h	C ₂₁ H ₂₁ ClN ₂ O ₅ S (448)	56	221	56.12	56.18	4.69	4.72	6.20	6.24	7.07	7.14

* Uncorrected

Table 8. Spectral Data of Compounds (4a-h)

Compd.	¹ H NMR (δ, ppm)							
	Ar-H	-OH	-CONH	-CH	-CH ₃	-OCH ₃	-OC ₂ H ₅	-OCH ₂ O-cyclic
4a	6.65-7.80 (m, 8H)	11.72 (s)	11.84 (s)	5.92 (s)	-	-	-	-
4b	6.63-7.78 (m, 8H)	11.80 (s)	11.90 (s)	5.95 (s)	-	3.86 (3H)(s)	-	-
4c	6.70-7.79 (m, 8H)	11.83 (s)	11.87 (s)	5.90 (s)	-	-	-	-
4d	6.65-7.80 (m, 8H)	11.79 (s)	11.85 (s)	5.92 (s)	-	-	-	-
4e	6.71-7.78 (m, 8H)	11.81 (s)	11.93 (s)	5.95 (s)	2.38 (3H) (s)	-	-	-
4f	6.64-7.76 (m, 7H)	11.84 (s)	11.98 (s)	5.91 (s)	-	-	-	6.08, (2H) (s)
4g	6.63-7.80 (m, 7H)	11.85 (s)	11.84 (s)	5.93 (s)	-	3.86 (3H) (s)	-	-
4h	6.64-7.79 (m, 7H)	11.81 (s)	11.96 (s)	5.95 (s)	-	-	4.06, 4H, (q) (CH ₂) 1.35, 6H, (t) (CH ₃)	-

An equimolar solution of 4-chloro-2-hydroxy-N-(4-oxo-2-aryl-thiazolidin-3-yl)benzamide (3a-h) and benzaldehyde in dioxane (50mL) in the presence of C₂H₅ONa were refluxed for about 3 h. The solvent was removed in vacuo. The resulting product was recrystallized from methanol to yield compound (4a-h).

The yields, melting points and other characterization data of these compounds are given in Table 7.

Table 9. Spectral Data of Compounds (4a–h)

Compd	13C NMR (δ , ppm)									
	Ar-H	-C-	-CONH	-CH	-CO	-CH ₂	-OCH ₃	-OC ₂ H ₅	-CH ₃	-OCH ₂ O- cyclic
4a	112.1-160.8	140.4	164.9	70.3	168.8	112.9	-	-	-	-
4b	112.7-160.5	140.3	164.7	70.3	168.9	113	56.0	-	-	-
4c	112.3-160.3	140.5	164.5	70.4	168.5	113	-	-	-	-
4d	113.0-160.8	140.5	165.2	70.2	168.7	112.8	-	-	-	-
4e	112.5-160.6	140.0	165.2	70.0	168.6	112.9	-	-	21.6	-
4f	112.6-159.8	140.4	164.8	70.1	168.9	112.6	-	-	-	101.8
4g	112.5-159.5	140.0	165.0	70.2	168.4	112.7	55.5	-	-	-
4h	112.8-160.8	140.3	164.9	70.4	169.0	112.5	-	65.2(OCH ₂) 15.1 (CH ₃)	-	-

Table 10. Antibacterial Activity of Compounds (3a–h)

Compounds	Gram +Ve		Gram –Ve	
	<i>Staphylococcus aureus</i>	<i>Bacillus subtilis</i>	<i>E. coli</i>	<i>Klebsiella promioe</i>
3a	12	12	11	16
3b	15	14	13	18
3c	11	15	17	15
3d	13	13	15	13
3e	16	12	16	12
3f	17	18	18	20
3g	13	13	16	15
3h	18	16	19	19
Tetracycline	22	19	21	23

Table 11. Antifungal Activity of Compounds (3a–h)

Compnds	Zone of Inhibition at 1000 ppm (%)				
	<i>Nigrospora sp.</i>	<i>Aspergillus niger</i>	<i>Botrydepl adia thibromi</i>	<i>Rhizopus nigricum</i>	<i>Fusarium oxyporium</i>
3a	65	62	66	60	68
3b	58	58	65	63	69
3c	67	67	67	67	71
3d	61	62	69	71	64
3e	60	64	63	72	69
3f	62	58	67	63	72
3g	64	69	70	60	68
3h	58	71	72	69	65

BIOLOGICAL SCREENING

Antibacterial activities

The antibacterial activities of all the compounds were studied against gram-positive bacteria (*Staphylococcus aureus* and *Bacillus subtilis*) and gram-negative bacteria (*E. coli* and *Klebsiella promioe*) at a concentration of 50 μ g/mL by agar cup plate method. A methanol system was used as control in this method. Similar conditions using tetracycline as a control was used standard for comparison. The area of inhibition of zone measured in cm. Compounds 3g, 3h, 4g, and 4h were found more toxic for microbes. Other

Table 12. Antibacterial Activity of Compounds (4a–h)

Compounds	Gram +Ve		Gram -Ve	
	<i>Staphylococcus aureus</i>	<i>Bacillus subtilis</i>	<i>E. coli</i>	<i>Klebsiella promioe</i>
4a	12	11	12	10
4b	15	10	11	13
4c	16	18	12	12
4d	10	13	14	15
4e	13	14	12	16
4f	14	12	13	12
4g	18	17	18	19
4h	18	16	19	19
Tetracycline	22	19	21	23

Table 13. Antifungal Activity of Compounds (4a–h)

Compnds	Zone of Inhibition at 1000 ppm (%)				
	<i>Nigrospora sp.</i>	<i>Aspergillus niger</i>	<i>Botrydepl adia thibromi</i>	<i>Rhizopus nigricum</i>	<i>Fusarium oxyporium</i>
4a	68	69	68	50	58
4b	66	61	62	63	65
4c	65	70	61	62	67
4d	61	61	61	68	66
4e	63	62	60	71	62
4f	64	55	67	60	68
4g	69	59	69	56	69
4h	61	64	71	72	65

compounds found to be less or moderate active than tetracycline Tables 10 and 12.

Antifungal Activities

The fungicidal activity of all the compounds was studied at 1000 ppm concentration in vitro. Plant pathogenic organisms used were *Nigrospora sp.*, *Aspergillus niger*, *Botrydepl adia thibromi*, and *Rhizopus nigricum*, *Fusarium oxyporium*. The antifungal activity of all the compounds (3a–h) & (4a–h) were measured on each of these plant pathogenic strains on a potato dextrose agar (PDA) medium. Such a PDA medium contained potato 200g, dextrose 20g, agar 20g and water 1c. Five days old cultures were employed. The compounds to be tested were suspended (1000ppm) in a PDA

medium and autoclaved at 120° C for 15 min. at 15 atm. pressure. These media were poured into sterile Petri plates and the organisms were inoculated after cooling the Petri plates. The percentage inhibition for fungi was calculated after five days using the formula given below:

$$\text{Percentage of inhibition} = 100(X-Y) / X$$

Where X = Area of colony in control plate

Y = Area of colony in test plate

The fungicidal activity displayed by various compounds (3a-h) and (4a-h) is shown in Tables-11 and 13.

RESULTS AND DISCUSSION

It was observed that 4-chloro-2-hydroxy benzoic acid hydrazide (1), on condensation with aromatic aldehydes, yields 4-chloro-2-hydroxy benzoic acid arylidene hydrazides (2a-h). The structures of (2a-h) were confirmed by elemental analysis and IR spectra showing an absorption band at 1620–1640 (C=N), 3030-3080 cm⁻¹ (C-H, of Ar.), 3240-3260 cm⁻¹ (-OH), 1635–1640 cm⁻¹ (C=O of CONH), 3350 – 3500 cm⁻¹ (- NH), 2815, 1250 cm⁻¹ (-OCH₃), 2950, 1370 cm⁻¹ (-CH₃). The C, H, N analysis data, ¹H NMR, and ¹³C NMR spectral data of all compounds are presented in Tables 1, 2 and 3 respectively.

The structures assigned to 4-chloro-2-hydroxy-N (4-oxo-2-aryl- thiazolidin -3-yl) - benzamides (3a-h) were supported by the elemental analysis and IR spectra showing an absorption bands at 1630–1650cm⁻¹ (C=O of thiazolidinone ring), 740 – 750 cm⁻¹ (C-S-C of thiazolidinone ring), 3075-3095cm⁻¹ (CH₂ of thiazolidinone ring), 3030-3080 cm⁻¹ (C-H, of Ar.), 3240-3260 cm⁻¹ (-OH), 1630 – 1640 cm⁻¹ (C=O of CONH), 3350 – 3500 cm⁻¹ (- NH). The C, H, N analysis data, ¹H NMR, and ¹³C NMR spectral data of all compounds are presented in Tables 4, 5 and 6 respectively.

The IR spectra of (4a-h) are almost resemble those of the corresponding (3a-h) only discernable difference observed that the new band (but not strong) at 1628cm⁻¹ (-C=CH-Ar) is observed in all the spectra of (4a-h) Which might be responsible.

The C, H, N analysis data, ¹H NMR, and ¹³C NMR spectral data of all compounds are presented in Tables 7, 8 and 9 respectively.

The examination of elemental analytical data reveals that the elemental contents are consistence with the predicted structure shown in Scheme-1. The IR data also direct for assignment of the predicted structure. The final structure of all

compounds is confirmed by LC-MS. LC-MS of 3d and 4f compounds are 378 and 407 respectively.

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СИНТЕЗ И БИОЛОГИЧНА АКТИВНОСТ НА 4-ХЛОРО-2-ХИДРОКСИ- N -(5-МЕТИЛЕНЕ-4-ОКСО-2-АРИЛ-ТИАЗОЛИДИН-3-ИЛ) БЕНЗАМИД

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(Резюме)

Хидразидът на 4-хлоро-2-хидрокси бензоената киселина (1) претърпява лесна кондензация с ароматни алдехиди, за да се получат съответните арилиден хидразици на 4-хлоро-2-хидрокси бензоената киселина (2a-h) с добри добиви. Циклокондензацията на съединения (2a-h) с тиогликолова киселина дава 4-хлоро-2-хидрокси-N (4-оксо-2-арил-тиазолидин-3-ил) бензамиди (3a-h). Тези съединения (3a-h) реагират с бензалдехид в присъствието на натриев етанолат с получаване на 4-хлоро-2-хидрокси-N (5-метилен-4-оксо-2-арил-тиазолидин-3-ил) бензамиди (4 a-h). Структурите на тези съединения са установени въз основа на аналитични и спектрални данни. Всички новосинтезирани съединения са оценени за тяхната антибактериална и противогъбична активност.