Surface phase detection of proton-exchanged layers in LiNbO₃ and LiTaO₃ by IR reflection spectroscopy

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The proton exchange (PE) technology has focused scientists' attention for the last 30 years because of the easy and fast obtaining of waveguides with strong waveguiding effect in electro-optical crystals like LiNbO₃ (LN) and LiTaO₃ (LT). The Li_{1-x}H_xNbO₃/Li_{1-x}H_xTaO₃ layer, formed by Li-H ion exchange, shows complex phase behavior depending on the hydrogen concentration (value of x). Up to seven phases exist in PE-LN an up to five – in PE-LT. Each phase forms its own sublayer in the protonated region, the one with the highest value of x being on the top. That is why the recognition of the top sublayer phase in many cases could be used for some conclusions about the phases building the rest of the whole layer or for the optical and electro-optical quality of the optical waveguiding layer. It has been established that IR reflection spectra of proton-exchanged layers contain new bands within the range 850–1050 cm⁻¹ and each phase has its own reflection spectrum. This way, IR reflection spectra in low-frequency range could determine the top-layer phase status. The samples investigated were obtained at different technological conditions and the analysis performed contributes to the accumulation of knowledge about the technological control of the phase composition of proton-exchanged waveguide layers in LN and LT.

Key words: LiNbO₃, LiTaO₃, optical waveguides, proton exchange, phase composition.

INTRODUCTION

Lithium niobate and lithium tantalate are among the most attractive ferroelectric crystals for integrated optics. Being an alternative of the most popular technology for obtaining optical waveguides in $LiNbO_3 - Ti$ -indiffusion, proton exchange (PE) [1, 2] has undergone a strong development in the last two decades. PE represents a chemical reaction (diffusion and Li-H ion exchange) which takes place in the surface layer of a crystal immersed in an appropriate melt.

Going by the scheme:

$$LiMO_3 + xH^+ \Rightarrow Li_{1-x}H_xMO_3 + xLi^+$$
 (M=Nb,Ta)

PE modifies the surface layer (several μ m in depth) by Li–H ion exchange causing a large extraordinary index change $\Delta n_e (\Delta n_e \cong 0.12 \text{ for LiNbO}_3)$ and $\cong 0.02$ for LiTaO₃ at 0.633 µm). The PE layers show complex phase behaviour depending on the hydrogen concentration (value of x) and causing significant decrease in electro-optical coefficients and increase in optical losses and instabilities.

The phase model developed by Korkishko et al. [3, 4], on which contemporary ideas on phase formation in monocrystalline layers of $\text{Li}_{1-x}\text{H}_x\text{MO}_3$ are based, suggests that each phase originates as an individual sublayer of several hundred nm or less. In every single one of them Δn_e is a linear function of concentration. The lattice parameters vary among phases. Within each phase, the change of the extraordinary refractive index Δn_e is proportional to x; within a phase transition, the value of Δn_e and/ or of the deformations perpendicular to the surface change by leap. Up to 7 phases could be formed in LiNbO₃ (α , κ_1 , κ_2 , β_1 , β_2 , β_3 , β_4), and up to 5 – in LiTaO₃ (α , κ , β , γ , δ).

The complicated phase composition of the proton-exchanged layers, determined by the degree of H^+ -Li⁺ substitution, has oriented the main efforts since the beginning of the technology to the ways for controlling the phase composition and to methods for waveguide phase characterization.

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Material	Sample	Proton source	T [°C]	t [h]	Ta [h]	t _a [h]	М	d [μm]	Δn_{e}	Possible phase composition	Surface phase
LiTaO ₃	TZ-2	LiHSO ₄ (vapors)	250	20	_	_	1	1.23	0.0162	α, κ, δ	δ
$L11aO_3$	1Z-3	$L1HSO_4$ (vapors)	220	48	-	_	1	1.44	0.0101	α, κ, ο	0
LiTaO ₃	TZ-4	LiHSO ₄ (vapors)	200	72	_	_	1	2.09	0.0058	α, κ, δ	δ
LiTaO ₃	T-1	Benzoic acid	240	8	265 295	1 1.5	2	2.18	0.0213	β(γ)	γ
LiTaO ₂	T-2	Benzoic acid	240	31	400	2	5	9.45	0.0155	А	α
LiTaO ₃	Z-1	Benzoic acid	240	8	295 350	1 1	4	4.25	0.0189	α, κ	α, κ
LiNbO ₃	LZ-1	LiHSO ₄ (vapors)	250	3.5	_	_	6	2.66	0.1491	$\beta_1, \beta_3, \beta_4$	β_{4}
LiNbO ₃	NM-1	NH_4HSO_4 (melt)	230	3.3	-	_	9	2.47	0.1508	$\beta_1, \beta_3, \beta_4$	β_4
LiNbO ₃	NM-5	LiHSO ₄ (melt)	175	1.5	-	—	1	0.62	0.1222	$\beta_1, \beta_2, \beta_3$	β_1

Table 1. Technological and waveguide parameters of the proton-exchanged waveguides: (T – temperature, t – duration of the PE process, T_a and t_a – temperature and duration of the annealing, M – number of waveguide modes at $\lambda = 633$ nm, Δn_e – extraordinary refractive index change, d – waveguide depth)

EXPERIMENTAL

Proton-exchanged optical waveguides in Z-cut LiNbO₃ and Z-cut LiTaO₃ were obtained by different technology conditions described in Table 1.

The phase analysis based on mode and IR absorption spectra was performed and reported earlier [5–8], the results being presented also in Table 1. In addition, infrared reflection spectra were registered in order to confirm the phase composition as well as to give some indications about the distribution of the different phases within the protonexchanged layers.

IR reflection spectra were recorded at angle of incidence $\theta = 70^{\circ}$ (measured from the normal to the surface) as shown in Fig. 1. The spectrometer accessories for specular reflectance were fixed- angle ones (for 20° and 70°). Since the penetration depth depends on the angle of incidence, at smaller angles (closer to the normal incidence) deeper penetration



Fig. 1. Schematic sketch of IR reflection measurements

takes place and the measured spectra are affected by the presence of the various phases forming the waveguide. It was established [9] that at 70° the spectrum of the surface layer is separated from those of deeper situated layers in multiphase guides. This way only the surface phase could contribute to the reflection spectra of multiphase waveguides and the IR reflection spectroscopy allows the surface phase to be recognized.

RESULTS AND DISCUSSION

LiNbO₃

The IR reflection spectra for LN samples are presented in Fig. 2.

The reflection IR spectra of protonated LN crystals have new bands compared to the spectra of virgin crystal (LZ-ref) or α -phase PE-LiNbO₃ [9]. These bands appear at frequences in the range of 890–1010 cm⁻¹, each phase having its own spectrum. It has been found that the new bands which can be attributed to new phases are 975 cm⁻¹ for β_1 phase, 980 cm⁻¹ for β_2 and β_3 phases, 970 cm⁻¹ for β_4 phase [9, 10]. The new band at 890–985 cm⁻¹ appears after PE in addition to the lattice spectrum of LN.

The low-frequency edges of the band correspond to the TO phonons of the NbO₆ vibration mode. Less distorted NbO₆ octahedra without any non-bridging oxygen ions are present in the α , κ_1 and κ_2 phases, as well as in pure LN. The appearance of weak extra bands in the region of IR reflection spectra from 800 to 900 cm⁻¹ for all β_i phases suggests the presence of more distorted NbO₆ octahedra with non-bridging oxygen ions. Each β_i phase is marked by a very specific lattice vibration spectrum with unique charac-



Fig. 2. IR reflection spectra measured at $\theta = 70^{\circ}$ for the LiNbO₃ proton exchanged layers

teristic bands. A significant difference between the crystalline structures of different β_i phases exists, which is introduced by the added chemical bonds with characteristic frequencies of vibration. This feature can be used for accurate identification of β_i phases in any $H_xLi_{1-x}NbO_3$ waveguide.

Since the analysis based on mode and IR absorption spectroscopy suggests a particular phase composition, where the most strongly protonated phase is at the surface of the waveguide, particular changes in the reflection spectrum can be attributed to the respective phase. This way the analysis was made which allows a particular frequency band to be attributed to the presence of the definite phase on the surface.

The new bands which can be attributed to new phases are: 955 cm⁻¹ and 970 cm⁻¹ for κ_2 -phase, 965 cm⁻¹ for κ_1 -phase and 965 cm⁻¹ for α -phase.

Thus, looking at the spectra in Fig. 2 we could conclude that β_1 phase is present on the top of sample NM-5 while β_4 phase forms the top of the waveguiding layers of samples NM-1 and LZ-1, which are really strongly protonated. Also, it is seen that the spectrum of NM-5 is closest to the shape of the virgin sample (LZ-ref), which suggests that the contribution of the α -phase is larger than in the case of the other two samples. It could be seen that LZ-1 and NM-1 have almost the same spectra, confirming their equal phase composition determined by other methods, as it can be seen in Table 1.

In the case of analysis of the phase composition of the most weakly protonated waveguide, the reflection spectra give particularly important information since they show that the waveguide is a single-phase one. As it is known from the phase model, the single-phase layers form the layer in such way that the value of x increases towards the surface (i. e. the most strongly protonated layer is at the surface). In our case the uppermost layer is of the β_1 phase and therefore, according to the phase analysis based on mode and IR spectroscopy, the possible β_2 and β_3 phases are not present in the protonated layer. (Another possibility is that the layer is slightly "buried" but the technological conditions do not suggest such a result).

LiTaO₃

The IR reflection spectra for LT samples are presented in Fig. 3 and Fig. 4.

The main changes in the reflection spectra introduced by proton exchange in Z-cut LiTaO₂ occur in the range of 850–1050 cm⁻¹. They were compared to the spectra of X-cut PE-LiTaO₃ given in [10] and some correlations with lattice deformations and reflection minimums were made in order to assign the spectral changes to a definite phase. According to [10], the changes observed at 899, 952 and 985 cm⁻¹ in IR-reflection spectra of all samples could be assigned to the β - and γ -phases, and the change at about 1000 cm⁻¹ to the δ -phase, respectively. Since in the case of the waveguides (Fig. 3) obtained by proton exchange in vapors (PEV) the second perturbation is much stronger, we should conclude that the surface phase of all investigated waveguides is δ as their spectra are almost identical.

The phase composition of the annealed samples suggests the presence of α - and κ -phases, so the new bands in their reflection spectra (Fig. 4) could be assigned to these two phases: 960 cm⁻¹ for the κ -phase, 974 cm⁻¹ for the α -phase, 891 for both α - and κ -phases. For the β - and γ -phases the



Fig. 3. Infrared reflection spectra measured at $\theta = 70^{\circ}$ for the LiTaO₃-waveguides obtained in LiHSO₄ vapors



Fig. 4. Infrared reflection spectra measured at $\theta = 70^{\circ}$ for the post-exchange annealed LiTaO₃-samples

band has three components at 890–915, 960 and 992 cm⁻¹, which are shifted to lower frequencies compared to the bands for the γ - and δ -phase of PEV waveguides.

The results of the IR-spectra analysis for the investigated LN and LT samples are shown in the rightmost two columns of Table 1. The information given by these spectra allows the determination to be made, which phase forms the uppermost layer of the waveguide. Thus, the information given by the analysis of the IR-reflection spectra allows us to be much more specific when determining which phases build the investigated waveguide layers.

CONCLUSIONS

• The phase composition of the entire PE-layers was determined by combined mode and IR absorption spectroscopy.

• IR reflection spectra of proton-exchanged layers contain new bands within the range 890–1010 cm⁻¹ and each phase has its own reflection spectrum. This way, IR reflection spectra in the low-frequency range could determine the top-layer phase status.

• The low-frequency edges of the band correspond to the TO phonons of the NbO₆/TaO₆ vibration mode. The appearance of weak extra bands in the region of IR reflection spectra from 800 to 900 cm⁻¹ suggests the presence of more distorted

NbO₆/TaO₆ octahedra with non-bridging oxygen ions. There is a significant difference between the crystalline structures of different phases, which is introduced by the added chemical bonds with characteristic frequencies of vibration.

• The phase-characteristic bands for LiNbO₃ are at:

965 cm⁻¹ for both α - and κ_1 -phase;

955 cm⁻¹ for κ_2 -phase;

975 cm⁻¹ for β_1 -phase;

980 cm⁻¹ for β_2 - and β_3 -phases;

970 cm⁻¹ for β_4 -phase.

• The new bands assigned to the definite phase in the case of LiTaO₃ are:

960 cm⁻¹ for the κ -phase;

974 cm⁻¹ for the α -phase;

890 cm⁻¹ for both α - and κ -phases;

890–915, 960 and 992 cm⁻¹ for β - and γ -phases, the components of the first being shifted to lower frequencies compared to the bands of the second;

1000 cm⁻¹ for the δ -phase.

• The presented results could contribute to the obtaining of waveguides with control of their phase composition and therefore of their optical and electro-optical properties.

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ОПРЕДЕЛЯНЕ НА ПОВЪРХНИННАТА ФАЗА НА ПРОТОННО-ОБМЕНЕНИ СЛОЕВЕ В LiNbO₃ И LiTaO₃ ЧРЕЗ ОТРАЖАТЕЛНА ИЧ СПЕКТРОСКОПИЯ

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(Резюме)

Протонният обмен (PE) е технология, която привлича вниманието на учените през последните 30 години поради лесното и бързо получаване на оптични вълноводи със силен вълноводен ефект в електрооптичните кристали LiNbO₃ (LN) и LiTaO₃ (LT). Слоят Li_{1-x}H_xNbO₃/Li_{1-x}H_xTaO₃, получен чрез Li–H йонен обмен, има сложен фазов състав, зависещ от концентрацията на водорода (стойността на x). До седем фази могат да съществуват в PE-LN и до пет – в PE-LT. Всяка фаза образува свой собствен подслой в протонираната област, като този с най-висока стойност на x се намира на повърхността. По тези причини от определянето на фазата на повърхинния слой в много случаи може да се направят изводи и за фазите, изграждащи целия слой или за оптичните и електрооптичните качества на вълноводния слой. Установено е, че ИЧ отражателни спектри на протонно-обменените слоеве съдържат нови ивици в областта 850–1050 сm⁻¹ и всяка фаза има свой специфичен отражателен спектър. По този начин по ИЧ отражателни спектри в нискочестотната област може да се определи повърхнинната фаза. Тъй като изследваните проби са получени при различни технологични условия, проведеният анализ допринася за натрупване на знания по технологичен контрол на фазовия състав на протонно-обменени слоеве в LN и LT.