

SUPPLEMENTARY DATA

IR study on the electrochemical reduction of nimesulide

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Table S1. Theoretical B3LYP/6-311++G(2df,2dp) frequencies and intensities for nimesulide radical anion **2a**

No	Theoretical data		
	$\nu_{\text{theor.}}^{\text{a}}$	A^{b}	Approximate description ^c
1.	3370	93	$\nu(\text{N-H})$
2.	3095	4	$\nu_{\text{PhA}}(\text{C-H})$
3.	3091	1	$\nu_{\text{PhA}}(\text{C-H})$
4.	3062	4	$\nu_{\text{PhA}}(\text{C-H})$
5.	3061	2	$\nu_{\text{PhB}}(\text{C-H})$
6.	3058	16	$\nu_{\text{PhB}}(\text{C-H})$
7.	3051	31	$\nu_{\text{PhB}}(\text{C-H})$
8.	3042	11	$\nu_{\text{PhB}}(\text{C-H})$
9.	3038	0	$\nu^{\text{as}}(\text{CH}_3)$
10.	3035	1	$\nu_{\text{PhB}}(\text{C-H})$
11.	3026	0	$\nu^{\text{as}}(\text{CH}_3)$
12.	2939	0	$\nu^{\text{s}}(\text{CH}_3)$
13.	1602	49	$\nu_{\text{PhB}}(\text{C=C})$
14.	1599	39	$\nu_{\text{PhB}}(\text{C=C})$
15.	1584	152	$\nu_{\text{PhA}}(\text{C=C})$
16.	1552	67	$\nu_{\text{PhA}}(\text{C=C})$
17.	1500	121	$\delta_{\text{PhB}}(\text{C-H})$
18.	1489	180	$\delta_{\text{PhA}}(\text{C-H})$
19.	1467	5	$\delta_{\text{PhB}}(\text{C-H})$

20.	1440	74	$\nu_{\text{PhA}}(\text{C}=\text{C}); \delta_{\text{PhA}}(\text{C}-\text{H})$
21.	1434	48	$\delta^{\text{as}}(\text{CH}_3)$
22.	1431	18	$\delta^{\text{as}}(\text{CH}_3)$
23.	1400	43	$\delta(\text{N}-\text{H})$
24.	1365	400	$\nu(\text{C}-\text{NO}_2)$
25.	1340	1	$\delta_{\text{PhB}}(\text{C}-\text{H})$
26.	1338	20	$\delta^{\text{s}}(\text{CH}_3)$
27.	1323	56	$\nu_{\text{PhA}}(\text{C}=\text{C}); \nu_{\text{PhB}}(\text{C}=\text{C})$
28.	1308	15	$\nu_{\text{PhA}}(\text{C}=\text{C}); \nu_{\text{PhB}}(\text{C}=\text{C})$
29.	1289	708	$\nu^{\text{as}}(\text{SO}_2)$
30.	1271	28	$\delta_{\text{PhA}}(\text{C}-\text{H})$
31.	1257	102	$\nu^{\text{as}}(\text{NO}_2)$
32.	1238	717	$\nu(\text{CO}-\text{C}), \delta_{\text{PhA}}(\text{C}-\text{H})$
33.	1221	123	$\nu(\text{CO}-\text{C}), \delta_{\text{PhA}}(\text{C}-\text{H})$
34.	1188	54	$\nu_{\text{PhB}}(\text{C}=\text{C})$
35.	1179	17	$\nu_{\text{PhA}}(\text{C}=\text{C}); \nu_{\text{PhB}}(\text{C}=\text{C})$
36.	1178	5	$\nu_{\text{PhB}}(\text{C}=\text{C})$
37.	1137	236	$\nu^{\text{s}}(\text{SO}_2)$
38.	1103	17	$\nu_{\text{PhB}}(\text{C}=\text{C})$
39. ^d	1055	160	$\nu^{\text{s}}(\text{NO}_2)$

^a Infrared frequencies [cm^{-1}] scaled by Eqn. (1); ^b Predicted intensities [$\text{km}\cdot\text{mol}^{-1}$]; ^c Vibrational modes: ν , stretching; δ , in-plane bending; s , symmetrical; as , asymmetrical; PhA, phenyl ring A; PhB, phenyl ring B; ^d Followed by 54 lower-frequency vibrations.