

**SUPPLEMENTARY DATA**

*Ab Initio* SCF study of the barrier to internal rotation in simple amides

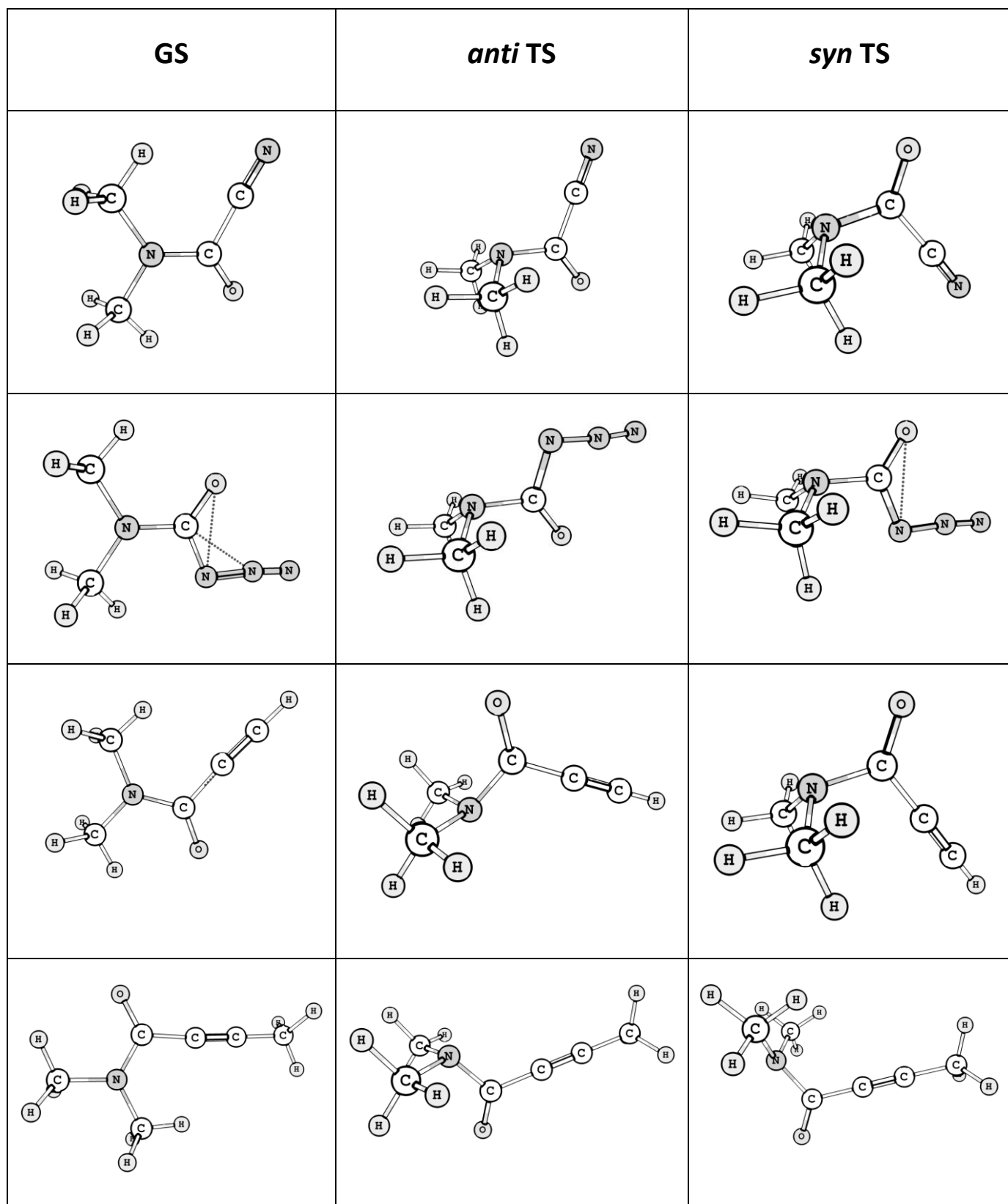
N. G. Vassilev

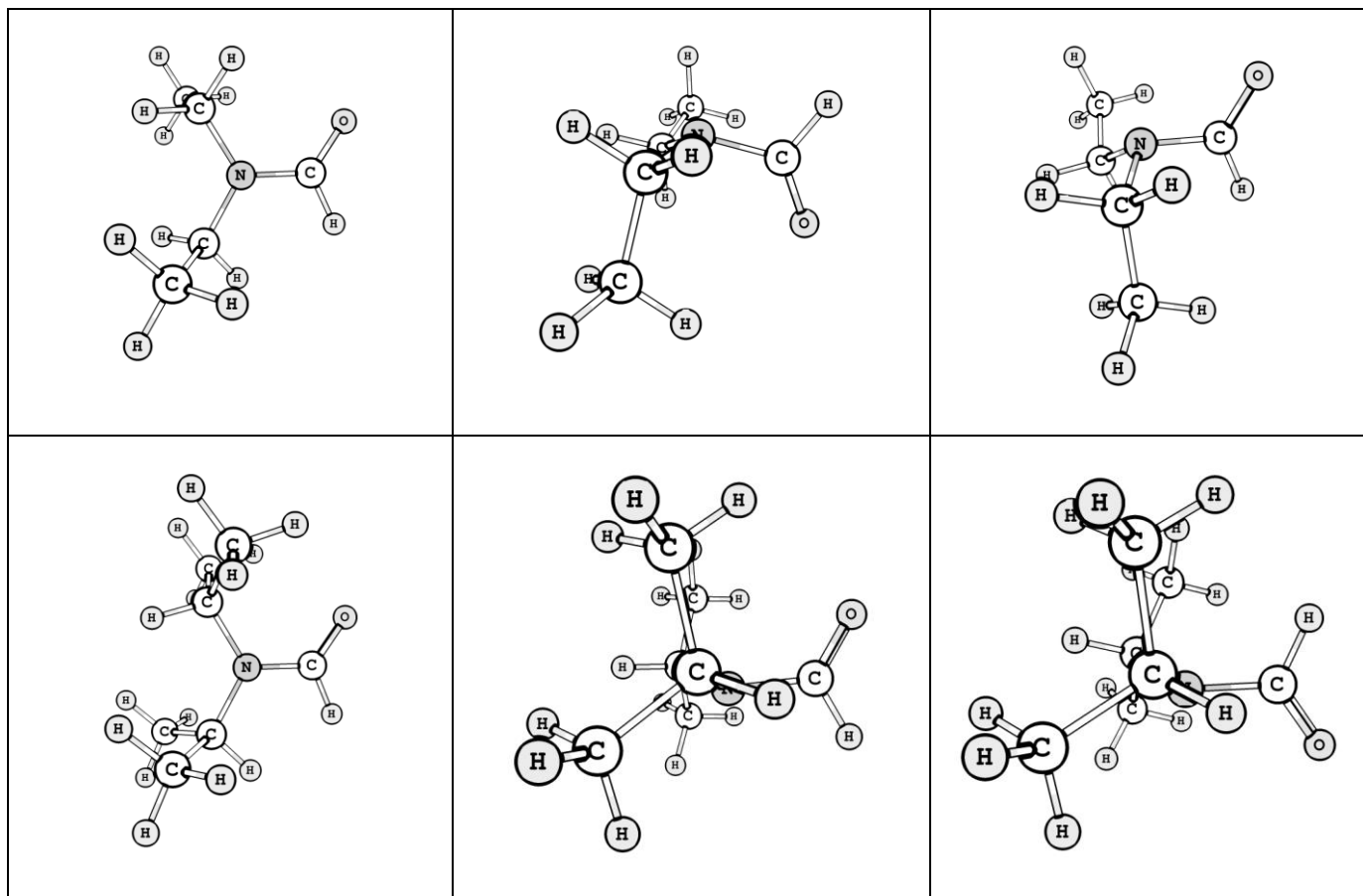
*Institute of Organic Chemistry with Centre of Phytochemistry, Bulgarian Academy of Sciences, Acad. G. Bonchev street, bl. 9, 1113 Sofia, Bulgaria*

e-mail: niki@orgchm.bas.bg

**Table of contents**

<b>Figure 1S.</b> Ground state (GS) and two transition states (TS) structures of the studied amides: (R'=CH <sub>3</sub> , R = CN, N <sub>3</sub> , C≡C-H, C≡C-CH <sub>3</sub> ; R'=H, R = C <sub>2</sub> H <sub>5</sub> , CH(CH <sub>3</sub> ) <sub>2</sub> )	2
<b>Table 1S.</b> Calculated energies for the studied amides R-C(O)NR' <sub>2</sub> in Hartrees	4
<b>Table 2S.</b> Origin of the difference in the rotational barriers (kcal/mol) of the studied amides R-CON(CH <sub>3</sub> ) <sub>2</sub> in the gas phase	5





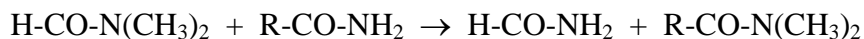
**Figure 1S.** Ground state (GS) and two transition states (TS) structures of the studied amides:  
 ( $R' = \text{CH}_3$ ,  $R = \text{CN}$ ,  $\text{N}_3$ ,  $\text{C}\equiv\text{C-H}$ ,  $\text{C}\equiv\text{C-CH}_3$ ;  $R' = \text{H}$ ,  $R = \text{C}_2\text{H}_5$ ,  $\text{CH}(\text{CH}_3)_2$ )

**Table 1S.** Calculated energies for the studied amides R-C(O)NR'<sub>2</sub> in Hartrees

R	R'	State	ZPE <sup>b</sup>	HF/6-31G*	MP2(fc)/6-31+G**//6-31G*	HF/6-311++G**	MP2(fc)/6-311++G**//6-311++G**	PCM/6-311++G**
CH <sub>3</sub> C≡C	CH <sub>3</sub>	GS	84.20	-361.705110	-362.822927	-361.800779	-363.035250	-7.2 <sup>c</sup>
CH <sub>3</sub> C≡C	CH <sub>3</sub>	<i>anti</i> TS	83.98	-361.680512 <sup>a</sup>	-362.796311 <sup>a</sup>	-361.776673 <sup>a</sup>	-363.009445 <sup>a</sup>	-5.6 <sup>c</sup>
CH <sub>3</sub> C≡C	CH <sub>3</sub>	<i>syn</i> TS	83.92	-361.678469	-362.795327	-361.774802	-363.009209	-6.6 <sup>c</sup>
HC≡C	CH <sub>3</sub>	GS	67.31	-322.657584	-323.645785	-322.744732	-323.828466	-6.9 <sup>c</sup>
HC≡C	CH <sub>3</sub>	<i>anti</i> TS	67.05	-322.632508 <sup>a</sup>	-323.618149 <sup>a</sup>	-322.720096 <sup>a</sup>	-323.801902 <sup>a</sup>	-5.3 <sup>c</sup>
HC≡C	CH <sub>3</sub>	<i>syn</i> TS	66.98	-322.630536	-323.617324	-322.718336	-323.801782	-6.1 <sup>c</sup>
N≡C	CH <sub>3</sub>	GS	60.93	-338.712595	-339.732431	-338.802302	-339.915725	-7.2 <sup>c</sup>
N≡C	CH <sub>3</sub>	<i>anti</i> TS	60.60	-338.685550 <sup>a</sup>	-339.701189	-338.775457 <sup>a</sup>	-339.885634	-5.4 <sup>c</sup>
N≡C	CH <sub>3</sub>	<i>syn</i> TS	60.48	-338.684050	-339.701347 <sup>a</sup>	-338.774331	-339.886375 <sup>a</sup>	-6.0 <sup>c</sup>
N <sub>3</sub>	CH <sub>3</sub>	GS	63.94	-409.694259	-410.926097	-409.803606	-411.140548	-3.4 <sup>d</sup>
N <sub>3</sub>	CH <sub>3</sub>	<i>anti</i> TS	63.70	-409.669832 <sup>a</sup>	-410.901043 <sup>a</sup>	-409.779826 <sup>a</sup>	-411.116520 <sup>a</sup>	-3.9 <sup>d</sup>
N <sub>3</sub>	CH <sub>3</sub>	<i>syn</i> TS	63.67	-409.667923	-410.900286	-409.778106	-411.116026	-4.0 <sup>d</sup>
H	C <sub>2</sub> H <sub>5</sub>	GS	107.70	-325.064127	-326.070618	-325.152833	-326.287625	-7.3 <sup>e</sup>
H	C <sub>2</sub> H <sub>5</sub>	<i>anti</i> TS	106.95	-325.031517 <sup>a</sup>	-326.035407 <sup>a</sup>	-325.120101 <sup>a</sup>	-326.253552 <sup>a</sup>	-4.4 <sup>e</sup>
H	C <sub>2</sub> H <sub>5</sub>	<i>syn</i> TS	106.73	-325.029234	-326.033911	-325.118689	-326.252694	-6.4 <sup>e</sup>
H	i-C <sub>3</sub> H <sub>7</sub>	GS	129.54	-403.131352	-404.411288	-403.238596	-404.688025	-3.7 <sup>f</sup>
H	i-C <sub>3</sub> H <sub>7</sub>	<i>anti</i> TS	128.60	-403.096938 <sup>a</sup>	-404.373029 <sup>a</sup>	-403.204072 <sup>a</sup>	-404.650673 <sup>a</sup>	-2.3 <sup>f</sup>
H	i-C <sub>3</sub> H <sub>7</sub>	<i>syn</i> TS	128.51	-403.095152	-404.371845	-403.203395	-404.650225	-3.3 <sup>f</sup>
CH <sub>3</sub> C≡C	H	GS	50.16	-283.649266	-284.500681	-283.732702	-284.658790	
CH <sub>3</sub> C≡C	H	<i>anti</i> TS	50.04	-283.624609 <sup>a</sup>	-284.475042 <sup>a</sup>	-283.707864 <sup>a</sup>	-284.634549 <sup>a</sup>	
CH <sub>3</sub> C≡C	H	<i>syn</i> TS	50.03	-283.621808	-284.472365	-283.705592	-284.632523	
HC≡C	H	GS	33.31	-244.601362	-245.323182	-244.676136	-245.451906	
HC≡C	H	<i>anti</i> TS	33.09	-244.576376 <sup>a</sup>	-245.296724 <sup>a</sup>	-244.651003 <sup>a</sup>	-245.426980 <sup>a</sup>	
HC≡C	H	<i>syn</i> TS	33.06	-244.573643	-245.294253	-244.648782	-245.425100	
N≡C	H	GS	26.98	-260.654219	-261.407503	-260.730905	-261.536917	
N≡C	H	<i>anti</i> TS	26.60	-260.628429 <sup>a</sup>	-261.379307 <sup>a</sup>	-260.705090 <sup>a</sup>	-261.510356 <sup>a</sup>	
N≡C	H	<i>syn</i> TS	26.52	-260.625624	-261.376991	-260.702800	-261.508455	
N <sub>3</sub>	H	GS	29.92	-331.638605	-332.604377	-331.735570	-332.764778	
N <sub>3</sub>	H	<i>anti</i> TS	29.80	-331.613499 <sup>a</sup>	-332.579125 <sup>a</sup>	-331.710547 <sup>a</sup>	-332.741088 <sup>a</sup>	
N <sub>3</sub>	H	<i>syn</i> TS	29.76	-331.612009	-332.577935	-331.709426	-332.740094	

<sup>a</sup> more stable TS; <sup>b</sup> ZPE is reported in kcal/mol at the HF/6-31G\* level scaled by 0.89; <sup>c</sup> Energy change from gas phase to TCE in kcal/mol; <sup>d</sup> Energy change from gas phase to CCl<sub>4</sub> in kcal/mol; <sup>e</sup> Energy change from gas phase to diethyl formamide in kcal/mol; <sup>f</sup> Energy change from gas phase to toluene in kcal/mol.

**Table 2S.** Origin of the difference in the rotational barriers (kcal/mol) of the studied amides  
R-CON(CH<sub>3</sub>)<sub>2</sub> in the gas phase



R	H	CN	N <sub>3</sub>	C≡CH	C≡CCH <sub>3</sub>
$\Delta\Delta\text{H}^0(\text{GS})$	0	0.9(0.4)	2.6(2.5)	2.3(2.2)	2.5(2.5)
$\Delta\Delta\text{H}^0(\text{anti TS})$	0	-1.2(-1.5)	-0.8(-1.0)	-0.6(-0.8)	-0.5(-0.6)
$\Delta\Delta\text{H}^0(\text{syn TS})$	0	-0.2(-0.6)	1.4(1.2)	0.8(0.6)	0.9(0.8)
<i>anti TS - GS:</i>					
$\Delta\Delta\text{H}^\ddagger(0\text{K})$	0	-0.4(-1.6)	-4.3(-1.6)	-2.6(-2.9)	-3.2(-3.2)
$\Delta\Delta\text{H}^0(\text{anti TS}) - \Delta\Delta\text{H}^0(\text{GS})$	0	-2.1(-1.9)	-3.4(-3.5)	-2.9(-3.0)	-3.0(-3.2)
<i>syn TS - GS:</i>					
$\Delta\Delta\text{H}^\ddagger(0\text{K})$	0	-0.8(-1.4)	-4.1(-1.4)	-2.3(-2.3)	-2.8(-2.5)
$\Delta\Delta\text{H}^0(\text{syn TS}) - \Delta\Delta\text{H}^0(\text{GS})$	0	-1.0(-0.9)	-1.3(-1.4)	-1.5(-1.5)	-1.7(-1.7)
$\Delta\Delta\text{G}_{\text{eff}}^\ddagger(298\text{K})$	0	-0.7(-1.9)	-3.3(-0.6)	-2.3(-2.5)	-2.3(-2.1)
$\Delta\Delta\text{G}_{\text{exp}}^\ddagger(298\text{K})$	0	-0.4(gas)	-2.9(gas)	-1.3(liquid)	-1.3(liquid)

$\Delta\Delta\text{H}^0(\text{GS})$ ,  $\Delta\Delta\text{H}^0(\text{anti TS})$  and  $\Delta\Delta\text{H}^0(\text{syn TS})$  are the energy changes (HF/6-31G\* energy plus ZPE correction) for the model reaction in the ground, *anti* and *syn* transition states, respectively.  $\Delta\Delta\text{H}^\ddagger(0\text{K})$  is the difference in the calculated enthalpy of activation for the amide rotation,  $\Delta\Delta\text{G}_{\text{eff}}^\ddagger(298\text{K})$  is the change in the calculated free energy and  $\Delta\Delta\text{G}_{\text{exp}}^\ddagger(298\text{K})$  is the experimental change in free energy. The values in parenthesis correspond to the HF/6-311++G\*\* level of theory.